

Jairo Quiroga\*, Angelina Hormaza and Braulio Insuasty

Department of Chemistry, Universidad del Valle, A. A. 25360 Cali, Colombia

Claudio Saitz[a], Carolina Jullian [b] and Alvaro Cañete [a]

[a] Departamento de Química Orgánica y Físico-Química,

[b] CEPEDEQ, Universidad de Chile, Casilla 233,

Santiago de Chile, Chile

Received May 27, 1997

A series of pyrazolo[1,5-*a*]pyrimidin-3-ones **3** was prepared from Meldrum's acid and 5-amino-3-arylpyrazoles **1** by cyclization in nitrobenzene of the corresponding 5-pyrazolylaminomethylene Meldrum's acid derivatives **2**. The structure of pyrazolo[1,5-*a*]pyrimidin-3-ones and their precursors were determined by nmr measurements.

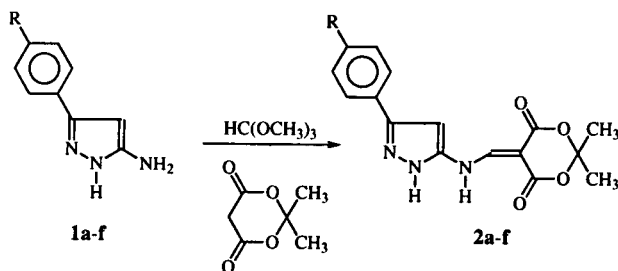
*J. Heterocyclic Chem.*, **35**, 61 (1998).

In the last twenty years considerable interest has been focused on derivatives of pyrazolo[1,5-*a*]pyrimidine, due to their physiological and biological activities [1-9]. In our previous work we have described some procedures for the synthesis of aromatic derivatives of pyrazolo[1,5-*a*]pyrimidine [10,11].

Continuing with the research on aminopyrazoles [10-13] in this work we studied the reaction of 5-amino-3-arylpyrazoles **1** with methoxymethylene derivatives of Meldrum's acid. A solution of Meldrum's acid and methyl orthoformate (1:5) was heated to reflux for 2.5 hours and immediately the 5-amino-3-arylpyrazole **1** was added in an equimolecular amount relative to Meldrum's acid. The reaction mixture was heated for 10-15 minutes, cooled and the precipitate which formed was filtered, to give the corresponding 5-pyrazolylaminomethylene derivative of Meldrum's acid **2**.

The structures of compounds **2a-f** were established using spectroscopic methods. Thus, the ir spectra of compounds **2a-f** measured in potassium bromide pellets show two bands for the elongation vibrations of C=O groups at 1680-1735 cm<sup>-1</sup> and two bands for the NH groups at 3150-3330 cm<sup>-1</sup>. The <sup>1</sup>H-nmr spectra of compounds **2**, measured in dimethyl-d<sub>6</sub> sulfoxide, besides the signal of the methylene groups at 1.6-1.7 ppm, showed: two doublets at δ 8.70-9.12 ppm and 11.32-11.40 ppm in a 1:1 relationship, corresponding to the =CH and NH protons of NH=CH group, two singlet for =CH and NH protons of the pyrazole ring at δ 6.90-7.15 and δ 13.20-13.68 ppm respectively and multiplet aromatic protons at δ 7.18-8.29 ppm.

Scheme 1



Compound	R	mp (°C)	Yield, %
<b>2a</b>	H	327-328	70
<b>2b</b>	CH <sub>3</sub>	341	80
<b>2c</b>	CH <sub>3</sub> O	245	48
<b>2d</b>	Cl	347	63
<b>2e</b>	Br	352-353	57
<b>2f</b>	NO <sub>2</sub>	342	57

Table 1

<sup>1</sup>H-NMR Data of **2** (δ values, TMS as the Internal Standard, in Dimethyl-d<sub>6</sub> Sulfoxide, 300 MHz)

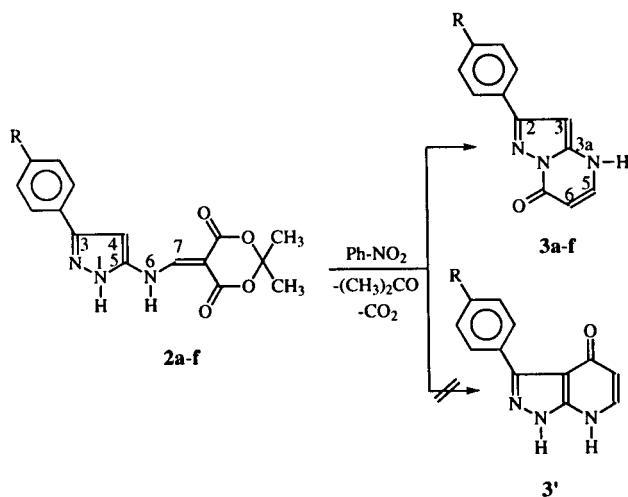
Compound	4-H s	7-H d	1-NH s	6-NH d	CH <sub>3</sub> s	Ar m
<b>2a</b>	6.90	8.72	13.30	11.32	1.70	7.58-7.77
<b>2b</b>	6.80	8.70	13.20	11.35	1.68	7.21-7.62
<b>2c</b>	6.79	8.69	13.14	11.28	1.67	7.02-7.68
<b>2d</b>	6.95	8.74	13.39	11.38	1.60	7.98-8.29
<b>2e</b>	6.91	8.77	13.41	11.37	1.70	7.73-7.75
<b>2f</b>	7.15	8.70	13.68	11.40	1.75	7.98-8.29

CH<sub>3</sub> for **2b** 2.32 and OCH<sub>3</sub> for **2c** 3.83 ppm.

The cyclization of the compounds **2** was carried out by heating to reflux in nitrobenzene (20% w/w) for 30 minutes to give compounds **3**. The structures of pyrazolo[1,5-*a*]pyrimidin-3-ones **3a-f** were established from their spectral characteristics.

In the ir spectra of compounds **3** measured in potassium bromide pellets a band for C=O group at 1670-1680  $\text{cm}^{-1}$  and a band for NH group at 3120-3145  $\text{cm}^{-1}$  were observed.

Scheme 2



Compound	R	mp (°C)	Yield, %
<b>3a</b>	H	364-365 [a]	48
<b>3b</b>	CH <sub>3</sub>	356	44
<b>3c</b>	CH <sub>3</sub> O	332-333	48
<b>3d</b>	Cl	378-379	45
<b>3e</b>	Br	375-376	46
<b>3f</b>	NO <sub>2</sub>	368	47

[a] Literature [9] mp 345-348 and [14] mp 303-306.

In the <sup>1</sup>H-nmr spectra of compounds **3** (Table 2) measured in dimethyl-d<sub>6</sub> sulfoxide the signal for the methyl groups disappeared, and two doublets at  $\delta$  7.06-7.93 and 5.70-5.81 ppm in a 1:1 ratio, corresponding to the CH<sub>(5)</sub>=CH<sub>(6)</sub>, fragment of pyrimidine ring can be observed together with a singlet for the =CH proton of pyrazole ring at  $\delta$  6.70-6.90 ppm. This evidence is to establish the reaction route **2** → **3a-f**, eliminating the formation of compounds **3'**.

Table 2

<sup>1</sup>H-NMR Data of **3** ( $\delta$  values, TMS as the Internal Standard, in Dimethyl-d<sub>6</sub> Sulfoxide, 300 MHz)

Compound	3-H s	5-H d	6-H d	NH s	Ar m
<b>3a</b>	6.69	7.41	5.73	12.39	7.41-8.00
<b>3b</b>	6.60	7.82	5.71	12.39	7.28-7.86
<b>3c</b>	6.58	7.04	5.71	12.41	7.82-7.96
<b>3d</b>	6.68	7.86	5.75	12.42	7.52-8.01
<b>3e</b>	6.71	7.80	5.72	12.40	7.62-7.90
<b>3f</b>	6.89	7.93	5.82	12.59	8.20-8.40

CH<sub>3</sub> for **3b** 2.36 and OCH<sub>3</sub> for **3c** 3.83 ppm.

The <sup>13</sup>C-nmr data for **3** are summarized in Table 3. Signal assignments were made based on DEPT experiments and data from previous work [11,15]. Significant features are as follows: the signal for C-3 appeared at  $\delta$  96.6-96.7 ppm, C-5 was observed at  $\delta$  143.9-153.7 ppm, C-6 registered at  $\delta$  86.5-88.3 ppm and C=O at  $\delta$  157.2-160.8 ppm.

Table 3

<sup>13</sup>C-NMR Data of **3** ( $\delta$  values, TMS as the Internal Standard, in Dimethyl-d<sub>6</sub> Sulfoxide, 300 MHz)

Compound	<b>3a</b>	<b>3b</b>	<b>3c</b>	<b>3d</b>	<b>3e</b>	<b>3f</b>
C-2	140.3	139.7	140.0	140.4	140.4	144.2
C-3	96.7	96.2	96.6	96.7	96.7	96.6
C-3a	153.8	153.4	153.7	152.6	152.7	151.6
C-5	143.9	143.4	143.8	143.9	143.9	148.3
C-6	87.0	86.4	86.5	87.2	87.2	88.3
C=O	157.3	156.9	157.3	157.2	157.2	157.2
Ar	C <sub>i</sub>	133.3	138.9	126.3	132.2	139.6
	C <sub>o,m</sub>	127.1	126.6	115.0	128.8	129.1
		129.6	129.8	128.4	129.7	132.6
C <sub>p</sub>	129.8	130.1	160.8	134.4	132.5	140.7

CH<sub>3</sub> for **3b** 2.14 and OCH<sub>3</sub> for **3c** 3.83 ppm.

## EXPERIMENTAL

Melting points were taken on a Büchi melting point apparatus and are uncorrected. The ir spectra were obtained in potassium bromide pellets with a Perkin-Elmer 599B spectrometer. The <sup>1</sup>H- and <sup>13</sup>C-nmr spectra were run on a Bruker DRX 300 spectrometer in dimethyl-d<sub>6</sub> sulfoxide. The mass spectra were recorded on a Fison MD-LC 800 (EI) operating at 70 eV. The elemental analysis have been obtained using a LEGO CHNS-900 equipment.

5-[3-(*p*-R-Phenylpyrazolylamino)methylene Meldrum's Acid Derivatives **2a-f**.

General Procedure.

A solution of Meldrum's acid (6.94 mmoles) and (34.7 mmoles) of methyl orthoformate was refluxed for 2.5 hours, then 6.94 mmoles of 5-amino-3-(*p*-R-phenyl)pyrazoles **1a-f** were added. The reaction mixture was heated for 10-15 minutes and the precipitate was filtered, to give the corresponding 5-[3-(*p*-R-phenylpyrazolylamino)methylene Meldrum's acid derivatives **2a-f**.

5-[3-Phenylpyrazolylamino)methylene Meldrum's Acid Derivative **2a**.

This compound was obtained *via* the general procedure as yellow crystals; <sup>13</sup>C-nmr (dimethyl-d<sub>6</sub> sulfoxide, ppm): 27.4 (2, CH<sub>3</sub>), 87.3 (=C(C=O)<sub>2</sub>), 93.7 (C-4), 105.1 (C(CH<sub>3</sub>)<sub>2</sub>), 152.8 (C-7), 163.7 (C=O), 164.6 (C=O); ms: (70 eV) m/z (%) 313 (14, M<sup>+</sup>), 255 (35), 238 (41), 237 (100), 211 (43), 159 (41), 149 (95), 130 (41), 77 (46).

Anal. Calcd. for C<sub>16</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>: C, 61.34; H, 4.83; N, 13.41. Found: C, 61.42; H, 4.78; N, 13.46.

5-[3-(*p*-Methylphenyl)pyrazolylamino]methylene Meldrum's Acid Derivative **2b**.

This compound was obtained according to general procedure as pale yellow crystals;  $^{13}\text{C}$ -nmr (dimethyl- $d_6$  sulfoxide, ppm): 21.7 (*p*-CH<sub>3</sub>), 27.3 (2, CH<sub>3</sub>), 87.2 (=C(C=O)<sub>2</sub>), 93.2 (C-4), 105.1 (C(CH<sub>3</sub>)<sub>2</sub>), 152.8 (C-7), 163.3 (C=O), 164.6 (C=O); ms: (70 eV) *m/z* (%) = 327 (5, M<sup>+</sup>), 252 (19), 251 (88), 177 (25), 173 (67), 149 (100), 71 (33).

*Anal.* Calcd. for C<sub>17</sub>H<sub>17</sub>N<sub>3</sub>O<sub>4</sub>: C, 62.38; H, 5.23, N, 12.84. Found: C, 62.48; H, 5.26; N, 12.80.

5-[3-(*p*-Methoxyphenyl)pyrazolylamino]methylene Meldrum's Acid Derivative **2c**.

This compound was obtained according to general procedure as pale yellow crystals;  $^{13}\text{C}$ -nmr (dimethyl- $d_6$  sulfoxide, ppm): 27.3 (2, CH<sub>3</sub>), 56.1 (OCH<sub>3</sub>), 87.2 (=C(C=O)<sub>2</sub>), 92.7 (C-4), 105.1 (C(CH<sub>3</sub>)<sub>2</sub>), 152.8 (C-7), 163.7 (C=O), 164.6 (C=O); ms: (70 eV) *m/z* (%) = 343 (28, M<sup>+</sup>), 268 (80), 267 (100), 252 (27), 241 (33), 213 (40), 198 (45), 160 (38), 134 (46), 89 (48), 77 (25).

*Anal.* Calcd. for C<sub>17</sub>H<sub>17</sub>N<sub>3</sub>O<sub>5</sub>: C, 59.47; H, 4.99; N, 12.24. Found: C, 59.53; H, 4.92; N, 12.19.

5-[3-(*p*-Chlorophenyl)pyrazolylamino]methylene Meldrum's Acid Derivative **2d**.

This compound was obtained according to general procedure as pale yellow crystals;  $^{13}\text{C}$ -nmr (dimethyl- $d_6$ , ppm): 27.4 (2, CH<sub>3</sub>), 87.3 (=C(C=O)<sub>2</sub>), 94.0 (C-4), 105.1 (C(CH<sub>3</sub>)<sub>2</sub>), 152.8 (C-7), 163.5 (C=O), 164.3 (C=O); ms: (70 eV) *m/z* (%) = 349/347 (2/6, M<sup>+</sup>), 289 (12), 273 (31), 271 (100), 182 (19), 149 (85), 71 (23).

*Anal.* Calcd. for C<sub>16</sub>H<sub>14</sub>N<sub>3</sub>O<sub>4</sub>Cl: C, 55.26; H, 4.06; N, 12.08. Found: C, 55.20; H, 4.12; N, 12.02.

5-[3-(*p*-Bromophenyl)pyrazolylamino]methylene Meldrum's Acid Derivative **2e**.

This compound was obtained according to general procedure as yellow crystals.  $^{13}\text{C}$ -nmr (dimethyl- $d_6$  sulfoxide, ppm): 27.4 (2, CH<sub>3</sub>), 87.4 (=C(C=O)<sub>2</sub>), 94.1 (C-4), 105.1 (C(CH<sub>3</sub>)<sub>2</sub>), 152.8 (C-7), 163.4 (C=O), 164.6 (C=O); ms: (70 eV) *m/z* (%) = 393/391 (3/3, M<sup>+</sup>), 318 (17), 317 (72), 315 (89), 239 (42), 149 (100), 71 (72).

*Anal.* Calcd. for C<sub>16</sub>H<sub>14</sub>N<sub>3</sub>O<sub>4</sub>Br: C, 49.00; H, 3.60; N, 10.71. Found: C, 49.06; H, 3.68; N, 10.77.

5-[3-(*p*-Nitrophenyl)pyrazolylamino]methylene Meldrum's Acid Derivative **2f**.

This compound was obtained according to general procedure as yellow crystals.  $^{13}\text{C}$ -nmr (dimethyl- $d_6$  sulfoxide, ppm): 27.4 (2, CH<sub>3</sub>), 87.5 (=C(C=O)<sub>2</sub>), 94.8 (C-4), 105.1 (C(CH<sub>3</sub>)<sub>2</sub>), 152.8 (C-7), 163.5 (C=O), 164.5 (C=O); ms: (70 eV) *m/z* (%) = 358 (2, M<sup>+</sup>), 282 (28), 205 (24), 204 (100), 174 (86), 158 (46), 149 (62), 145 (67), 77 (45).

*Anal.* Calcd. for C<sub>16</sub>H<sub>14</sub>N<sub>4</sub>O<sub>6</sub>: C, 53.63; H, 3.94; N, 15.64. Found: C, 53.59; H, 3.98; N, 15.58.

Cyclization of 6-[2-R-3-R<sub>1</sub>-3,4-Dihydro-4-oxopyrimidinyl-amino]methylene Meldrum's Acid Derivatives **2a-f**.

## General Procedure.

Compounds **2a-f** (1 mmole) in nitrobenzene (20% w/w) was heated to reflux for 30 minutes. The cyclized products (**3a-f**) were isolated by cooling, followed by filtration, washing with ethanol, and drying.

2-Phenyl-4,7-dihydropyrazolo[2,3-*d*]pyrimidin-7-one **3a**.

This compound was obtained according to general procedure as white crystals; ms: (70 eV) *m/z* (%) = 211 (72, M<sup>+</sup>), 177 (23), 154 (24), 149 (100), 97 (27), 71 (47).

*Anal.* Calcd. for C<sub>12</sub>H<sub>9</sub>N<sub>3</sub>O: C, 68.24; H, 4.29; N, 19.89. Found: C, 68.29; H, 4.33; N, 19.82.

2-(*p*-Methylphenyl)-4,7-dihydropyrazolo[2,3-*d*]pyrimidin-7-one **3b**.

This compound was obtained according to general procedure as pale yellow crystals; ms: (70 eV) *m/z* (%) = 225 (100, M<sup>+</sup>), 197 (25), 168 (22), 154 (38), 115 (66), 91 (18), 89 (21), 71 (15).

*Anal.* Calcd. for C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>O: C, 69.32; H, 4.92; N, 18.65. Found: C, 69.38; H, 4.95; N, 18.60.

2-(*p*-Methoxyphenyl)-4,7-dihydropyrazolo[2,3-*d*]pyrimidin-7-one **3c**.

This compound was obtained according to general procedure as pale yellow crystals; ms: (70 eV) *m/z* (%) = 241 (93, M<sup>+</sup>), 198 (67), 149 (100), 77 (25), 71 (34).

*Anal.* Calcd. for C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>O<sub>2</sub>: C, 64.72; H, 4.60; N, 17.42. Found: C, 64.78; H, 4.55; N, 17.47.

2-(*p*-Chlorophenyl)-4,7-dihydropyrazolo[2,3-*d*]pyrimidin-7-one **3d**.

This compound was obtained according to general procedure as pale yellow crystals; ms: (70 eV) *m/z* (%) = 247/245 (18/52, M<sup>+</sup>), 177 (24), 154 (22), 149 (100), 97 (44), 71 (51).

*Anal.* Calcd. for C<sub>12</sub>H<sub>8</sub>N<sub>3</sub>OCl: C, 58.67; H, 3.28; N, 17.10. Found: C, 58.75; H, 3.33; N, 17.18.

2-(*p*-Bromophenyl)-4,7-dihydropyrazolo[2,3-*d*]pyrimidin-7-one **3e**.

This compound was obtained according to general procedure as yellow crystals; ms: (70 eV) *m/z* (%) = 291/289 (50/48, M<sup>+</sup>), 182 (17), 154 (33), 149 (100), 81 (34), 71 (42).

*Anal.* Calcd. for C<sub>12</sub>H<sub>8</sub>N<sub>3</sub>OBr: C, 49.68; H, 2.78; N, 14.48. Found: C, 49.73; H, 2.81; N, 14.44.

2-(*p*-Nitrophenyl)-4,7-dihydropyrazolo[2,3-*d*]pyrimidin-7-one **3f**.

This compound was obtained according to general procedure as yellow crystals; ms: (70 eV) *m/z* (%) = 256 (10, M<sup>+</sup>), 224 (18), 149 (100), 129 (29), 71 (33).

*Anal.* Calcd. for C<sub>12</sub>H<sub>8</sub>N<sub>4</sub>O<sub>3</sub>: C, 56.25; H, 3.15; N, 21.87. Found: C, 56.21; H, 3.10; N, 21.91.

## Acknowledgment.

We are grateful to COLCIENCIAS and Universidad del Valle for financial support.

## REFERENCES AND NOTES

- [1] B. M. Lynch, M. A. Khan, S. C. Schama and H. C. Teo, *Can. J. Chem.*, **53**, 119 (1975).
- [2] R. K. Robins, D. E. O'Brien, T. Novinson and H. Springer, *German Offen.* 2,257,547; *Chem. Abstr.*, **79**, 78840 (1973).
- [3] T. Novinson, R. K. Robins and D. E. O'Brien, *J. Heterocyclic Chem.*, **10**, 887 (1973).
- [4] A. A. Albert, R. K. Robins and D. E. O'Brien, *J. Heterocyclic Chem.*, **10**, 885 (1973).

- [5] M. H. Elnagdi, S. M. Fahmy, E. M. Zayed and M. A. M. Ilias, *Z. Naturforsch.*, **31b**, 795 (1976) and references cited therein.
- [6] M. H. Elnagdi, M. R. H. Elmoghayar and G. E. H. Elgemeie, *Synthesis*, **1** (1984).
- [7] J. V. Greenhill in *Comprehensive Heterocyclic Chemistry*, Vol 5, A. R. Katritzky and C. W. Rees, eds, 1984, p 305.
- [8] M. H. Elnagdi, M. R. H. Elmoghayar and G. E. Elgemeie, *Adv. Heterocyclic Chem.*, **41**, 319 (1984).
- [9] K. Senga, T. Novison and H. R. Wilson, *J. Med. Chem.*, **24**, 610 (1981).
- [10] V. D. Orlov, J. Quiroga, N. N. Kolos and S. M. Desenko, *Khim. Geterosikl. Soedin.*, 962 (1988).
- [11] J. Quiroga, B. Insuasty, R. Rincón, M. Larrahondo, N. Hanold and H. Meier, *J. Heterocyclic Chem.*, **31**, 1333 (1994).
- [12] V. D. Orlov, J. Quiroga and N. N. Kolos, *Khim. Geterosikl. Soedin.*, 1247 (1987).
- [13] J. Quiroga, B. Insuasty, M. Marin, A. Aguirre and H. Meier, *Rev. Col. Quim.*, **21**, 29 (1992).
- [14] H. Reimlinger, M. A. Peiren and R. Merenyi, *Chem. Ber.*, **103**, 3252 (1970).
- [15] S. Chimichi and B. Cosimelli, *Can. J. Chem.*, **70**, 1093 (1992).